M.Sc 4th Semester Examination, 2011 CHEMISTRY

PAPER — CH - 2204

The figures in the right-hand margin indicate marks

(Inorganic Special)

[Environmental Chemistry]

Full Marks: 40

Time: 2 hours

Answer any four questions

- 1. (a) What are the important factors that affect sampling?
 - (b) How high-volume sampler works for sampling of particulates?
 - (c) Write an account on monitoring of SO₂ in air.

3

2.	(a) Describe the method for the estimation of t following parameters in water samples:	the 4 + 4
	(i) Fluoride	
	(ii) B.O.D.	
	(b) How mercury is analyzed by AAS?	2
3.	(a) Name the pre-concentration technique involved before analysis of the material during sampling and explain them.	
	(b) How hollow cathode lamp works? Which g is used in ICPES?	gas 3 + 1
	(c) What is the basic difference between GSC a GLC?	ind 1
4.	(a) Schematically explain the working principle gas chromatography.	of 4
	(b) The Flame Ionization Detector is most use for which kind of sample?	eful 1
	(c) What are the advantages of graphite furnace	e? 3
	(d) How ICPES is superior than other atomiz devices?	ing 2

5.	(a)	What is neutron activation analysis? Discuss its		
		advantages. 1 +	3	
	(b)	Give a block diagram of FTIR spectrometer.	4	
	(c)	How can internal combustion engines be		
		modified to make autoexhausts free from		
		pollutants?	2	
6.	(a)	Draw the conventional flow diagram of		
		municipal wastewater treatment plant.	3	
	(<i>b</i>)	Discuss oxygen sag curve in detail.	4	
	(c)	Write the merits and demerits of the trickling filters with activated sludge system in		
		wastewater treatment process.	3	
7.	. (a)	Discuss the method of removal of NO, from		
	•	a coal fired power plant flue gas.	5	
	(b)	Write a note on metal ion toxicity.	5	

(Organic and Physical Special)

[Polymer]

Full Marks: 40

Time: 2 hours

Answer any four questions

- (a) Give an average composition of natural rubber latex. Why preservation of natural rubber latex is necessary? How is it done?
 - (b) Describe the manufacture of smoked sheet rubber from natural rubber latex. 5
- 2. (a) What do you mean by compounding of rubber?

 Name the main compounding ingredients used in the compounding of rubber giving suitable example in each case.

 2+4
 - (b) Write a short note on vulcanisation of rubber. 4
- 3. What is nitrile rubber? Discuss in brief its preparation, properties and uses. 2+4+2+2

4.	Draw a neat sketch of an extruder and hence describe	
	the extrusion process of a polymer.	10

- 5. (a) What do you mean by phenolic resins?

 Describe the manufacturing process of resol mentioning thereby the reactions involved. How is resol converted to resite?

 1 + 4 + 2
 - (b) What do you mean by environmental stress cracking of polythene?
- 6. (a) Describe the suspension polymerisation process for the manufacture of polyvinyl chloride.
 - (b) Discuss the effect of plasticizers in PVC.
- 7. (a) Name the raw materials used for the synthesis of epoxy resins. Write down the reaction leading to the formation of epoxy resin. 2+2
 - (b) How chlorosulphonated polyethylene (hypalon) is produced from low density polyethylene (LDPE)? Mention the main applications of hypalon. 2+2
 - (c) Write down the properties of isotactic polypropylene.

2

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- 3. Write short notes on any four of the following: $2\frac{1}{2}\times 4$
 - (i) Properties and uses of polystyrene
 - (ii) Compression moulding
 - (iii) Synthesis of polyhexamethylene terephthalate
 - (iv) Viscose rayon
 - (v) Synthesis of EPDM
 - (vi) Properties of stereoregular polyethylene.